A highly stereocontrolled total synthesis of dysiherbaine[†]

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A total synthesis of dysiherbaine, a potent agonist of AMPA-KA type glutamate receptors, has been accomplished in completely stereocontrolled manner starting from tri-*O*-acetyl-D-galactal in 25 steps and in 3% overall yield.

Dysiherbaine (1) and neodysiherbaine A (2) (Fig. 1), characteristic neuroexcitatory amino acids isolated from Micronesian sponge Dysidea herbacea.^{1,2} are known to be subunit-selective agonists of AMPA-KA type glutamate receptors. These compounds exhibit the most potent convulsant activities among the excitatory amino acids known to date such as kainic acid and domoic acid.³ Due to the low availability from natural sources as well as the intriguing biological activity and synthetically challenging molecular architectures, these compounds have attracted much attention in the chemical and biological communities.⁴ Thus, there have been a number of synthetic studies including total syntheses of dysiherbaine (1) and neodysiherbaine A (2).5-8 However, most of the syntheses are not efficient enough to obtain sufficient quantities of these natural products as well as their analogs, due to the lack of efficiency in the assembly of the cis-fused hexahydrofuro[3,2b]pyran ring system containing four contiguous all-cis stereogenic centres and one quaternary centre. Recently, in the synthesis of neodysiherbaine A (2),^{7c} we successfully developed an efficient methodology for the stereocontrolled construction of the abovementioned core structure. We herein report a new efficient synthesis of dyshiherbaine (1) based on the strategy we have established in the synthesis of neodysiherbaine A (2).



Fig. 1 Dysiherbaine and neodyshiherbaine A.

The synthetic route employed for the synthesis of **2** allowed us to envisage the retrosynthetic route consisting of 5-*exo-tet* cyclisation of epoxide **3** and cross-coupling⁹ of organozinc compound **5** with alkenyl iodide **6**, accessible stereoselectively by reductive iodination¹⁰ of **7**, as major transformations (Scheme 1). For the synthesis of **7**, we envisaged Donohoe's tethered intramolecular aminohydroxylation¹¹ of **8**, which was expected

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to be appropriate for the stereo- and regioselective construction of the three contiguous *cis* vicinal amino alcohol functionalities.‡



Scheme 1 Retrosynthetic analysis.

Our synthesis of 1 thus began with the preparation of allylic alcohol 12 (Scheme 2). Upon triethylsilane-promoted reduction, methanolysis, bis-silylation, and selective desilylation using NH_4F , tri-*O*-acetyl-D-galactal (9) was converted to alcohol 10 in good yield. According to Kotsuki's procedure,¹² 10 was triflated and the resulting triflate was then directly reacted with the lithium acetylide derived from propargyl tetrahydropyranyl ether to give 11. Desilylation of 11 with TBAF furnished the required allylic alcohol 12 quantitatively.

After conversion of **12** to carbamate **13** by the addition of trichloroacetyl isocyanate, its tethered aminohydroxylation was investigated according to the procedures reported by Donohoe and co-workers.¹¹ Upon treatment of **13** with *tert*-butyl hypochlorite, NaOH, $K_2OsO_2(OH)_2$ and *i*-Pr₂NEt in aqueous *n*-propanol at room temperature, cyclic carbamate **14** and **15** were stereoselectively produced in 57 and 8% yields, respectively, together with recovered **13** (9%) (Scheme 3). Although the initially produced **14** isomerised partially to **15** through equilibration as mentioned above, this isomerisation was not a serious problem

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Scheme 2 Reagents and conditions. (i) Et₃SiH, BF₃·OEt₂, CH₂Cl₂, 0 °C; (ii) NaOMe, MeOH, rt; (iii) *t*-BuMe₂SiCl, imidazole, DMF, rt; (iv) NH₄F, MeOH, rt; (v) 2,6-lutidine, Tf₂O, CH₂Cl₂, -78 °C; (vi) *n*-BuLi, THPOCH₂CCH, DMPU, THF, -65 °C; (vii) *n*-Bu₄NF (TBAF), THF, rt; (viii) CCl₃(CO)NCO, CH₂Cl₂, 0 °C, then K₂CO₃, MeOH, rt; (ix) NaOH, *t*-BuOCl (1.06 eq.), *i*-Pr₂NEt (0.048 eq.), K₂OsO₂(OH)₄ (0.1 eq.), *n*-PrOH, rt; (x) 2-methoxypropene, PPTS (0.07 eq.), DMF–acetone (27:7), 0 °C; (xi) LiAlH₄, THF, 0 °C; (xi) conc. HCl, MeOH, reflux, then NaOH, Boc₂O, rt; (xii) Ac₂O, pyridine; (xiii) 2,6-lutidine, Et₃SiOTf, CH₂Cl₂, -78 °C; (xiv) K₂CO₃, MeOH, rt; (xv) NaH₂Al(OCH₂CH₂OMe)₂ (Red-Al), Et₂O, 0 °C, then I₂, -40 °C to rt; (xvi) Et₃SiCl, Et₃N, DMAP (0.1 eq.), CH₂Cl₂, rt.

because the mixture of 14 and 15 thus obtained was converted to a single advanced intermediate 16 by a three-step sequence involving temporary protection as the acetonide,§ LiAlH₄ reduction and acidic hydrolysis followed by *tert*-butoxycarbonylation. By successive acetylation, silylation and methanolysis, the secondary hydroxyl groups of 16 were protected as their triethylsilyl ethers to give 17. Reaction¹⁰ of 17 with Red-Al, followed by treatment of the resulting alkenylaluminate complex with iodine, allowed stereo- and regioselective formation of the corresponding (*Z*)-iodoalkene, which was then protected as its triethylsilyl ether to give 18.

Having obtained key intermediate **18**, the final stage of our total synthesis was executed according to the methodology^{7c} we have previously established. Thus, alkenyl iodide **18** was subjected to palladium-catalysed cross-coupling reaction with organozinc reagent **5**, generated from (*R*)-*N*-Boc-3-iodoalanine methyl ester, to furnish coupling product **19** very cleanly. Without purification, deprotection of **19** using TBAF in the presence of acetic acid gave alcohol **20** in excellent yield. Katsuki–Sharpless asymmetric epoxidation¹³ using a catalytic amount of Ti(O-*i*-Pr)₄ and



Scheme 3 Reagents and conditions. (i) 5, $(PPh_3)_4Pd$ (0.2 eq.), benzene, HMPA, dimethylacetamide, 80 °C; (ii) TBAF (0.9 eq.), AcOH (1.2 eq.), THF, rt; (iii) diisopropyl L-tartrate (0.1 eq.), Ti(O-*i*-Pr)₄ (0.09 eq.), *t*-BuOOH (2 eq.), 4A molecular sieves, CH_2Cl_2 , -35 °C; (iv) TBAF (3 eq.), AcOH (4 eq.), THF, rt; (v) PPTS (1 eq.), CH_2Cl_2 , rt; (vi) NaIO₄, THF-H₂O, rt; (vii) *n*-Pr₄NRuO₄ (TPAP) (0.2 eq.), *N*-methylmorpholine *N*-oxide (NMO) (2 eq.), 4A molecular sieves, MeCN, rt; (viii) 6 M HCl, 80 °C, neutralisation with NaOH.

diisopropyl L-tartrate to afford epoxide **20** in perfect diastereoselectivity. In this epoxidation, the other diastereoisomeric epoxide was not detected. After removal of the triethylsilyl ether protecting groups of **20** using TBAF and acetic acid, the resulting triol **21** was successively subjected to acid-catalysed 5-*exo-tet* cyclisation, oxidative cleavage of the 1,2-glycol functionality and TPAP oxidation¹⁴ to yield lactam **23**. In this transformation, the corresponding C4 epimer was not obtained at all, suggesting that the cyclisation occurred with complete inversion of stereochemistry at the quaternary centre. Finally, upon acidic hydrolysis and neutralisation with NaOH, **23** furnished dysiherbaine (1) which was purified by reverse-phase HPLC. The synthetic substance was identical with natural dysiherbaine by spectroscopic (¹H and ¹³C NMR)¶ and chromatographic (reverse-phase TLC and HPLC) comparisons.

In conclusion, we have achieved an efficient synthesis of dysiherbaine (1) from tri-*O*-acetyl-D-galactal (9), which enables us to obtain this natural amino acid in sufficient quantities. The crucial assembly of the *cis*-fused hexahydrofuro[3,2-*b*]pyran ring system containing four contiguous all-*cis* asymmetric centres and one quaternary centre was accomplished in completely stereocontrolled manner by the combination of Donohoe's tethered aminohydroxylation and our previously developed methodology involving palladium-catalysed cross-coupling and Katsuki–Sharpless epoxidation followed by 5-*exo-tet* cyclisation. The synthetic method is of general value in approaches to related neuroexcitatory amino acids.

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Notes and references

 \ddagger Just recently, Cohen and Chamberlin⁶ reported an efficient synthesis of the tetrahydropyran core based on Donohoe's tethered aminohydroxylation; however, they have not synthesised dyshiherbaine.

§ Without this protection, the next LiAlH₄ reduction turned out to be unsuccessful to afford a complex mixture.

¶ In ¹H NMR (D₂O), chemical shifts of NCH₃ and CHNMe varied markedly depending on the concentration as we reported previously.^{5a} However, the ¹H NMR (D₂O) spectrum of synthetic dysiherbaine hydrochloride was found to be completely identical with that of natural specimen.

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